

1 **Impact of Heterotrophically Stressed Algae for Biofuel**
2 **Production via Hydrothermal Liquefaction and Catalytic**
3 **Hydrotreating in Continuous-Flow Reactors**

4
5 Karl O. Albrecht^{1,*}, Yunhua Zhu¹, Andrew J. Schmidt¹, Justin M. Billing¹, Todd R. Hart¹,
6 Susanne B. Jones¹, Gary Maupin¹, Richard Hallen¹, Toby Ahrens², Daniel Anderson¹

7
8 ¹Pacific Northwest National Laboratory, Energy & Environment Directorate. P.O. Box 999,
9 MSIN P8-60, Richland, Washington 99352

10 ²BioProcess Algae, LLC, 450 Regency Parkway, Suite 400, Omaha, Nebraska, 68114

11
12 **Keywords:** *hydrothermal liquefaction; catalytic hydrotreating; heterotrophically stressed algae;*
13 *techno-economic analysis*

14
15 *Corresponding author

16 Karl Albrecht

17 Pacific Northwest National Laboratory, P.O. Box 999, MSIN P8-60, Richland, WA 99352

18 e-mail: karl.albrecht@pnnl.gov

19 Phone: 1-509-371-3775

20 Fax: 1-509-372-4732

21 **Abstract**

22 Two algal feedstocks were prepared for direct comparison of their properties when
23 converted to liquid hydrocarbon fuel. The first feedstock was prepared by growing an algal
24 strain phototrophically using a bio-film based approach. The second feedstock employed the
25 same algal strain but was stressed heterotrophically to significantly increase the lipid
26 concentration. The algal feedstocks were converted to liquid hydrocarbon fuels. First, the whole
27 algae (i.e. not defatted or lipid extracted) were converted to an intermediate biocrude using
28 continuous hydrothermal liquefaction (HTL) at 350°C and 3000 psig. The biocrudes were
29 subsequently upgraded via catalytic hydrotreating (HT) at 400°C and 1500 psig to remove
30 oxygen and nitrogen as well as increase the hydrogen-to-carbon ratio. The yield and
31 composition of the products from HTL and HT processing of the feedstocks are compared. A
32 techno-economic analysis of the process for converting each feedstock to liquid fuels was also
33 conducted. The capital and operating costs associated with converting the feedstocks to finished
34 transportation fuels are reported. A fuel minimum selling price is presented as a function of the
35 cost of the algal feedstock delivered to the HTL conversion plant. Heterotrophic stressing of the
36 algae significantly increased the concentration of lipids compared to the phototrophically grown
37 algae. The high lipid concentration resulted in a doubling of the yield to biocrude, and hence
38 diesel fuel blendstock. Although heterotrophic stressing of algae is costly, results presented in
39 this study suggest that the significant increase in fuel yield over phototrophic growth could more
40 than offset increased feedstock production costs.

41

42 **1. Introduction**

43 Hydrothermal liquefaction (HTL) is a means for producing liquid hydrocarbon fuels from
44 wet feedstocks.[1] HTL involves processing biomass feedstocks in hot subcritical water under
45 sufficient pressure to keep the water in the condensed phase. HTL produces an intermediate
46 biocrude organic phase, which is gravity separable from the water fed with the biomass
47 feedstock to the process. Algal feedstocks are especially suited for HTL because of the potential
48 for high oil yields relative to other types of biomass. The carbon acyl chains of the lipid fraction
49 within algal biomass remain intact during HTL processing, resulting in near 100% recovery of
50 the lipid fraction of the algal feedstock. The carbon yield to fuel from algae is also higher for
51 HTL processing than lipid extraction techniques because much of the carbohydrate and protein
52 portions of the algae are converted and included in the biocrude intermediate. Algal biocrude
53 can be upgraded via catalytic hydrotreating (HT). Catalytic hydrotreating removes heteroatoms
54 such as oxygen and nitrogen while increasing the hydrogen-to-carbon ratio of the organic
55 product. The HT hydrocarbon product may be suitable as a fuel blendstock or a standalone fuel
56 after further processing (e.g. isomerization) to meet all fuel specifications.

57 Numerous accounts reporting batch HTL processing of algal feedstocks are available,
58 including several reviews.[2-4] However, relatively few publications are available related to
59 HTL conversion in continuous flow equipment. Recently, Elliott *et al.*[5] published a review
60 focused on continuous flow HTL processing and key differences between continuous and batch
61 HTL. A key difference between batch and continuous processing is the need for the former to
62 use solvents for biocrude product recovery. While continuous processing does not preclude the

63 use of solvents for separation, gravity separation can be employed in continuous processing to
64 separate biocrude from the aqueous phase.

65 Jazwari *et al.*[6] reported continuous flow HTL data for processing *Chlorella* and
66 *Spirulina* under a variety of conditions. Higher concentrations of solids in the feed slurry (10
67 wt% vs. 1 wt%) and higher temperature (350°C vs. 300°C) resulted in greater yields of biocrude.
68 Biocrudes obtained in continuous mode with residence times of 3-5 min at 275-300°C were
69 comparable to biocrudes produced in a batch reactor at 350°C with 60 min of residence time,
70 demonstrating that continuous processing can be employed to significantly improve throughput
71 rates with a continuous plug-flow system. A maximum 42 wt% biocrude yield (collected via
72 solvent extraction) was achieved at 350°C at 3 min of residence time within the reactor using a
73 10 wt% *Chlorella* slurry feedstock. The *Chlorella* algae only had 4 wt% lipids, demonstrating
74 again the potential for HTL to convert a large portion of the non-lipid fraction to biocrude for
75 subsequent upgrading and inclusion in hydrocarbon fuel blendstocks.

76 Elliott *et al.*[1] reported continuous flow results for HTL and subsequent biocrude
77 upgrading of four *Nannochloropsis* feedstocks. Slurries of 17-34 wt% solids were HTL
78 processed continuously at nominally 350°C and 3000 psi. Mass yields of biocrude were 38-64
79 wt% on a dry/ash free basis. Two of the feedstocks tested were the same *Nannochloropsis*
80 *oceanica* strain grown and harvested by Cellana, Inc. under different conditions. One version
81 was harvested after high growth operation (designated AGLL for “low lipid”) and another after
82 stressed, low-growth conditions (designated AGHL for “high lipid”). The AGLL and AGHL
83 strains reported by Elliott *et al.*[1] are the same strains characterized in the final report of the
84 National Alliance for Advanced Biofuels and Bio-Products (NAABB). In the NAABB report,
85 the AGLL strain is designated “KA19 Stressed (Low Lipid)” and the AGHL strain as “KA19

86 Stressed (High Lipid).”[7] These algal feedstocks had total lipid contents of 20.8% (Low Lipid)
87 and 36.1% (High Lipid), respectively. Interestingly, Elliott *et al.*[1] reported similar mass yields
88 of biocrude at 60.8 wt% and 63.6 wt% for the AGLL and AGHL feedstocks, respectively.
89 Ultimate analysis of the biocrudes also demonstrated similar carbon, hydrogen and oxygen
90 concentrations. The similar yield and composition of the biocrudes produced by HTL processing
91 of the same algal strains grown under dissimilar conditions with varying lipid fractions again
92 demonstrates the versatility and effectiveness of HTL. A large portion of the non-lipid fraction
93 of the algal feedstocks was converted and included with the biocrude for subsequent upgrading.
94 The AGLL (Low Lipid) biocrude was successfully catalytically hydrotreated and processed to
95 produce both jet fuel and diesel fuel.[8] The AGLL (Low Lipid) biocrude only required
96 hydrodeoxygenation (i.e. catalytic HT) and isomerization but not cracking to make on-spec jet
97 fuel. Other algal-derived feedstocks comprised of lipid extracts required hydrodeoxygenation,
98 isomerization and cracking to make on-spec jet fuel. The biocrude may not have required
99 cracking because of the presence of lighter, more volatile constituents in the hydrotreated
100 product, which are from the converted carbohydrate and protein fractions of algae not present in
101 lipid extracted algae oils.

102 Several accounts of techno-economic analyses (TEAs) of HTL and HT algae processing
103 have been published recently. Some reports have been published based on extrapolations of
104 HTL batch data to continuous models. Delrue *et al.*[9] found that diesel produced by whole
105 algae HTL and catalytic HT was about 4 times greater than the cost to produce petroleum diesel
106 on a constant energy basis. Part of the reason for the significant cost was due to the use of
107 photobioreactors (PBRs) instead of open pond/raceways. Diesel fuel produced from whole algae
108 HTL was found to be 12% less than fuel produced by lipid extraction, but about 25% more than

109 fuel produced by first lipid extracting algae and then further converting the defatted algae via
110 HTL. Ou *et al.*[10] found that fuel produced by HTL and HT of defatted (i.e. lipid extracted)
111 algal biomass could be competitive with petroleum-derived fuels. The process was modeled
112 with a large stirred tank HTL reactor as opposed to a presumably lower volume plug flow
113 system. The cost of algal feedstock was modeled at cost similar to wet distillers grains utilized
114 as livestock feed. Feedstock cost was found to be a major cost driver in the overall minimum
115 fuel selling price (MFSP); only product fuel yield was found to have a greater effect on MFSP.
116 A Monte-Carlo analysis employing the optimistic wet distillers grains feedstock cost assumption
117 determined the MFSP would fall between about \$2.30/gal and \$3.15/gal.

118 Zhu *et al.*[11] reported economics for HTL of lipid extracted algae based on continuous
119 flow HTL reactor data. A shell-and-tube HTL reactor was modeled with 51.2% yield to
120 biocrude. The results were similar to the work by Ou *et al.*[10] in that feedstock cost and
121 product yield were major cost drivers for the ultimate MFSP. Zhu *et al.*[11] determined the
122 upgrading (i.e. catalytic HT) equipment cost was another significant cost that affected the
123 ultimate MFSP. The MFSP was estimated to be between \$2.07 and \$7.11.

124 Jones *et al.*[12] reported a design case based on data generated from continuous HTL
125 processing of whole *Nannochloropsis* and *Chlorella* algae with subsequent catalytic HT. The
126 product yield of algae to HTL biocrude was modeled at 51 wt%. The MFSP was determined to
127 be \$4.77/gal for diesel for a 1340 U.S. tons /day plant. For this scenario, the feedstock was
128 modeled to cost \$430/ton, which constituted 74% of the diesel production cost.

129 Davis *et al.*[13] combined spatiotemporal algal growth data with a HTL conversion
130 process to generate data for a TEA and life-cycle analysis (LCA) to determine greenhouse gas
131 (GHG) reduction potential. HTL with catalytic HT was employed as the conversion process to

132 convert whole algae to liquid fuels. In contrast to previous studies, seasonal variation growth
133 variation of algae could be factored into the price. When accounting for seasonal variability, the
134 MFSP for diesel fuel was between \$10.7/gal and \$14.1/gal. GHG emission reduction was
135 negatively affected during winter operation. However, an increase in the MFSP resulted if
136 winter operation was halted. For example, the weighted average for the MFSP of diesel from
137 several gulf coast locations increased from \$11.0/gal to \$13.3/gal when winter operation was
138 omitted due to low growth conditions. However, operating in winter caused the GHG emissions
139 from the algal growth and fuel production process to exceed that of petroleum diesel during the
140 winter season even as operation of the pond lowered the fuel selling price.

141 The goal of this study is to directly compare the two algal feedstocks converted to fuels
142 using HTL and HT. The feedstocks are an identical strain grown via two different methods:
143 phototrophically versus heterotrophically. Heterotrophic cultivation of algae produces
144 significantly higher lipid concentrations within the algae compared to phototrophically grown
145 algae. The higher lipid concentration should improve the overall yield to biocrude and
146 ultimately upgraded hydrocarbon (i.e. fuel) product. The yields of the HTL and catalytic HT
147 processes for each algal feedstock and the composition of the products were measured and
148 compared. Other consideration such as H₂ consumption during HT and the fraction of
149 hydrocarbon product suitable for various fuels (e.g. naptha vs. diesel) are compared and
150 contrasted in the context of the starting algal feedstock. As the cultivation conditions and
151 associated costs of the algal feedstocks tested were proprietary, a TEA was conducted with focus
152 on the cost of the thermochemical conversion from whole algae to fuel. The cost of algae
153 production is presented as a sensitivity analysis wherein the MFSP of fuel from each algal

154 feedstock is presented as a function of the cost of the algal feedstock delivered to the HTL
155 conversion plant.

156 **2. Materials and Methods**

157 The algal biomass, equipment and procedures used for HTL processing and biocrude
158 upgrading along with the method used in the resulting TEA are described below.

159 *2.1 Algal Feedstock*

160 Two algal feedstocks were produced by BioProcess Algae, LLC (BPA) for HTL
161 processing. The algal feedstocks were comprised of unialgal (>98%) *Chlorella* cultures grown
162 in proprietary modifications of freshwater nutrient media. The algal feedstocks are hereafter
163 referred to as, "standard lipid" and "high lipid" feedstocks. The "standard lipid" feedstock,
164 designated SL, is a green algal strain grown phototrophically in greenhouses. The greenhouses
165 are integrated with an adjacent corn-ethanol biorefinery for sources of waste CO₂ and waste heat
166 The "high lipid" feedstock, designated HL, is the same algal strain, but was heterotrophically
167 stressed using sugar. Glucose was added to the media of heterotrophically stressed cultures.

168 Typical biochemical composition ranges for the algal feedstocks are provided in

169

170 Table 1. During the heterotrophic stressing most of the protein is consumed while the lipid
171 content more than doubled. The carbohydrate content remained qualitatively similar. The
172 elevated lipid and diminished protein contents of the HL algae compared to the SL algae
173 increased the concentration of the carbon and hydrogen at the expense of the oxygen and
174 nitrogen. Heterotrophic stressing also diminished the concentration of sulfur and phosphorous,
175 which was also likely correlated with the lower protein concentration in the HL algae.

176 Images of the feedstocks are shown in Figure 1. The SL algal feedstock is a green paste
177 while the HL feedstock is chlorotic.

178



179

180 **Figure 1.** SL algal feedstock (left – 14.7 wt% solids) and HL algal feedstock (right – 20.3 wt%
181 solids) algal slurry feed for HTL

182

183
 184 **Table 1.** Dry basis (DB) elemental and biochemical composition analyses of the SL and HL
 185 algal feedstocks.

	SL	HL
Elemental Analysis, wt% (DB)		
Carbon	47.7	63.3
Hydrogen	6.9	9.3
Nitrogen	8.7	0.5
Oxygen	29.2	22.2
Sulfur*	0.5	0.1
Phosphorous*	2.1	1.0
Ash	9.2	4.9
Total	101.6	100.1
Biochemical Composition Analysis, wt% (DB)		
Protein	27-59	3.5-4.1
Lipids	12-30	57-64
Carbohydrate	18-43	28-33
Ash	9	5

186 *Sulfur and phosphorous are assumed to report to the ash and are hence not included in the
 187 elemental balance total.

188 2.2 HTL Processing

189 Elliott *et al.*[1] provided a process flow diagram and detailed description of the
 190 continuous-flow HTL reactor system used in this testing. The HTL system configured for this
 191 work utilized a continuous stirred tank reactor (CSTR) with a working volume of 415 ml,
 192 followed by plug flow reactors (PFR) (270 ml total volume) in series. Liquid hourly space
 193 velocity (LHSV) calculations are based on the total volume of the CSTR and PFR. Steady state
 194 samples were collected for each feed over a 2.0 h operating window. Steady state was achieved
 195 by collecting three consecutive 40 min set-aside samples prior to starting the steady state period.
 196 The total time pumping feed to the system was about 4.25 hr total for each feed. . Other
 197 specific operating parameters of the SL and HL feedstocks utilized in the HTL processing are

198 reported in Table 2. Previous testing has established baseline continuous HTL conditions at
 199 nominally 350°C and 3000 psig to allow direct comparison of a wide range of algae species and
 200 remove any variability caused by the changes in processing conditions. These conditions are
 201 generally applied to not just algae but all biomass feedstocks, including wood, corn stover,
 202 switch grass, etc.[5, 14]

203

204 **Table 2.** HTL conditions employed during the liquefaction processing of the SL and HL algal
 205 feedstocks.

	SL	HL
Total dry solids in feed, wt%	14.7	20.3
Ash in sfeed, wt%	1.4	1.0
Average feed density, g/ml @ 20°C	1.042	1.015
Feed pH	6.62	4.84
Feed chemical oxygen demand, mg O/L	202,000	468,000
Feed processing rate, L/h	1.5	1.5
LHSV, L feed/L reactor/h	2.2	2.2
HTL reactor temperature, °C	349	350
HTL reactor pressure, psig	2990	2980
Biocrude generated during steady state operation, g	154	420

206

207 The concentration of dry solids in the feed was based on a qualitative assessment by
 208 process operators regarding the pumpability of the feedstock when preparing the slurry for
 209 processing. The HL feedstock was determined to be pumpable with less water; the SL feedstock
 210 required more water for processing, and thus has lower total dry solids in the feed

211 During an HTL run, the liquid product was collected in parallel sample holders. A
 212 sample holder could be valved out in order to collect liquid products while the system was
 213 online. HTL liquid products were drained from the collectors into separatory funnels. A lighter

214 oil (designated as biocrude) and heavier aqueous phase spontaneously separated and could be
215 readily separated by removing the bottom water layer from the oil via the stopcock in the
216 separatory funnel. The biocrude was not further dried after separation from the aqueous layer.
217 Elemental analysis was performed on the feed, separated oil, aqueous byproduct, and solids
218 collected in the filter housing to determine mass and elemental balances. CHN, O, and S were
219 analyzed by ASTM methods D5291, D5373, and D4239, respectively, by ALS Environmental
220 (Tucson, AZ). Trace element analysis was conducted via ICP-OES as described previously[1].
221 The Total Acid Number (TAN) was performed following ASTM D3339 and the moisture via
222 Karl Fisher titration following a modified version of ASTM D5530. The aqueous phase was
223 analyzed for chemical oxygen demand (COD) and pH, with spot checks for ammonia and trace
224 metals. Gas samples could be withdrawn manually from the vent line and analyzed every 30 to
225 60 min via gas chromatography (GC) using a method also described earlier.[1]

226 2.3 *Catalytic Hydrotreating (HT) of HTL Biocrude*

227 The phase-separated biocrude products were hydrotreated in a continuous flow
228 hydrotreater system, which is described by Elliott *et al.*[15] The hydrotreater was configured as
229 a single stage reactor with one heating zone. The catalyst was procured from Alfa Aesar
230 (Product No. 45579). The catalyst was supplied as extrudates and reported by the supplier to
231 typically consist of 3.4-4.5% cobalt oxide and 11.5-14.5% molybdenum oxide on alumina. 40.0
232 ml of catalyst ground and sieved to -30/+60 mesh was measured out using a graduated cylinder
233 with an ID similar to the reactor. The mass of sized catalyst loaded was 24.60 g. On top of the
234 40.0 ml catalyst bed, 1.28 g of the catalyst was loaded as unground extrudates to serve as a pre-

235 heating zone. The approximate volume of the extrudates serving as the pre-heating zone was 3.4
236 ml.

237 Prior to introducing biocrude, the catalyst was sulfided with 35 wt% di-tertbutyl disulfide
238 (DTBDS) in decane. The DTBDS and decane were obtained from Sigma Aldrich. First, the
239 reactor was pressurized to nominally 1500 psig with H₂. Once at pressure, the H₂ flow was
240 allowed to stabilize at 170 SCCM, whereupon the reactor was heated to 150°C. Once the reactor
241 reached 150°C, the DTBDS/decane solution was introduced with a flow rate of 0.120 ml/min
242 (volumetric flow rate determined at 21°C). The temperature was then ramped at 1.4°C/min to
243 400°C and soaked for 4 hr. The biocrude feed was initiated to the reactor immediately upon the
244 conclusion of sulfiding.

245 Reactor conditions during biocrude hydrotreating were nominally 400°C and 1500 psig.
246 The volumetric feed rate of both biocrudes was 0.133 mL/min at 40°C. The dry basis mass flow
247 rate of the HL biocrude was 0.119 g/min. The dry basis mass flow rate of the SL biocrude was
248 0.116 g/min. H₂ was co-fed to the reactor at 188 SCCM for both feeds. The HL biocrude was
249 processed first. The feed was changed to the SL biocrude after rinsing the feed pump and lines
250 with acetone and drying with air. A transition sample was collected after switching biocrude
251 feeds. Data reported for the HL biocrude is based on a steady state sample collected between
252 38.3-44.5 hr time-on-stream (TOS). The SL biocrude data is based on a steady state sample
253 collected between 49.7 and 55.5 hr TOS.

254 The bio-oils and hydrotreated organic and aqueous products were analyzed by ALS
255 Environmental (Tucson, AZ) for CHN (ASM D5373/D5291), O (modified ASTM D5373), S
256 (ASTM D4239/D1552), moisture via Karl Fisher analysis (ASTM D6869) and acid number (mg
257 KOH/g; ASM D3339). The sulfur concentration in the organic hydrotreated products (ppm

258 level) was determined at PNNL following ASTM D5453. Simulated distillation of the organic
259 products was also conducted following ASTM D2887.

260 2.4 *Techno-economic Analysis Methodology*

261 The approach to developing techno-economics for a conversion process is similar to that
262 employed in previous conceptual design reports.[11, 12, 14] The process models are developed
263 using AspenPlus process flow simulation software. The heat and material balances from the
264 simulations are used to estimate capital and operating costs. A factor of 30% is applied to the
265 total installed capital cost to arrive at the total capital investment used in the minimum fuel
266 conversion price calculation. The minimum fuel conversion price (the production cost
267 excluding feedstock costs where the net present value is zero for a given internal rate of return) is
268 calculated using discounted cash flow analysis. An “nth” plant design is assumed, reflecting a
269 future time when the technology is mature and several plants have already been built and
270 operated. Thus, the costs do not include “first-of-a-kind” plant costs, e.g., special financing,
271 equipment redundancies, large contingencies, longer startup times necessary for the first few
272 plants and low on-line availability.

273 **3. Results and Discussion**

274 The SL and HL algal slurries were liquefied at nominally 350°C and 3000 psig via
275 continuous HTL processing. The collected liquid product included an organic biocrude and an
276 aqueous phase, which were gravity separable. The phase separated biocrude was subsequently
277 hydrotreated at nominally 400°C and 1500 psig with a sulfided CoMo catalyst in order to
278 significantly reduce the amount of oxygen and nitrogen and increase the hydrogen-to-carbon

279 ratio. Hydrotreating of the black opaque biocrude produced a slightly-yellow transparent organic
280 fraction and a water-white aqueous fraction. The hydrotreater product was also phase separated
281 and subjected to various individual analyses.

282 3.1 *Hydrothermal Liquefaction*

283 A single HTL campaign was conducted to process both the SL and HL algal slurries.
284 The HTL process was brought to steady state with the HL feedstock, which took approximately 3
285 hours. Over the course of the next 2 hours, 3 samples were collected, and gravity phase
286 separated. After the HL feedstock had been processed, feeding of the SL feedstock commenced
287 and 2 hours of processing was allowed to transition from the HL to the SL feedstock. After the
288 feedstock transition period, the SL feedstock was processed for an additional 2 hours at steady
289 state with 3 samples collected, phase separated and analyzed.

290 The overall mass and elemental balances achieved during HTL processing of the SL and
291 HL slurries are presented in Table 3. The overall mass balance including the biocrude, aqueous
292 and gas products of the process at steady state was excellent for both the SL and HL feedstocks.
293 Elemental balances were generally within +/-10% of closure with the exception of sulfur and
294 phosphorous for the SL feedstock and nitrogen for the HL feedstock. The elemental balances
295 outside of the 10% window represent low concentration elements in the algal feedstocks that can
296 report to multiple phases, causing even small experimental errors to result in significant overall
297 deviations from balance closure.

298 The mass and carbon yields to the desired organic biocrude phase were significantly
299 greater for the HL feed compared to the SL feed (Table 3). The doubling of the biocrude yield
300 observed with the HL algae feedstock is due to the high concentration of lipids present from

301 heterotrophic growth. Lipids such as tri-acyl glycerides (TAGs) are known to undergo
302 hydrolysis reactions under hydrothermal conditions, which cleave acyl chains off of the glycerol
303 backbone, resulting in free fatty acids (FFAs).[1] Phospholipids and sulfolipids also contain
304 similar fatty acid chains, which also undergo hydrolysis. Fatty acid amides are also observed,
305 which may be from a secondary reaction of FFAs with ammonia. The hydrocarbon chain and
306 terminal acid/amide functionalities survive the HTL process. Owing to the increased
307 concentration of lipid, the HL algae feedstock produced significantly more organic biocrude
308 phase than the SL feedstock. Furthermore, with less protein in the HL algae, fewer low
309 molecular weight compounds containing nitrogen were produced. Previous studies of algal HTL
310 biocrude and aqueous phases determined that low carbon number (less than 15 carbon atoms)
311 compounds with one nitrogen atom present and low double bond equivalents reported
312 preferentially to the aqueous phase.[16] With the low nitrogen content of the HL feedstock,
313 fewer nitrogen containing compounds were formed, lowering the carbon yield to the aqueous
314 phase. The increased hydrophobicity of the lipids in the HL feedstock also resulted in less
315 oxygenates in the HTL aqueous phase. Data presented in Table 4 reveals a 30-40% decrease in
316 the total carbon, total organic carbon and chemical oxygen demand of the HL HTL aqueous
317 phase compared to the SL aqueous phase.

318

319

320

321

322

323 **Table 3.** HTL processing mass and elemental balances with normalized mass and carbon yields
 324 to the biocrude, aqueous, solid and gas fractions.

	SL	HL
Overall HTL Mass Balance, %		
	100	99
Elemental Balances, %		
Carbon	90	93
Hydrogen	93	92
Oxygen	102	101
Nitrogen	93	67
Sulfur	54	106
Phosphorus	122	99
Mass Yield (normalized), %		
Biocrude	35	71
Solids	1	1
Gas	9	4
Aqueous	55	25
Carbon Yield (normalized), %		
Biocrude	59	87
Solids	2.3	1.3
Gas	5.1	1.7
Aqueous	34	10

325

326 While the carbon yield to the biocrude phase is greater when using the heterotrophically
 327 stressed HL feedstock, the concentration of carbon in the SL and HL biocrude products was
 328 similar (Table 4). However, the atomic hydrogen-to-carbon (H:C) molar ratio is greater in the
 329 HL biocrude. The greater H:C molar ratio and lower nitrogen contents in the HL biocrude
 330 combine to result in diminished H₂ consumption requirements when the biocrude is further
 331 upgraded via catalytic HT. On the other hand, the oxygen content is greater in the HL biocrude.
 332 From the perspective of the subsequent HT step, oxygen is preferred over nitrogen in the
 333 biocrude as the stoichiometric H₂ demand is less than with oxygen. Nitrogen reduction requires

334 1.5 equivalents of H₂ to produce ammonia while 1.0 equivalents of H₂ are required to reduce an
335 equivalent of oxygen to water.

336 The carbon, oxygen and hydrogen concentrations of the HL biocrude correspond well
337 with the high lipid nature of the material. It is instructive to consider oleic acid as a model
338 compound which represents the overall composition of the HTL biocrude. Oleic acid has
339 carbon, oxygen, and hydrogen concentrations of 77 wt%, 11wt%, and 12 wt%, respectively.
340 These concentrations correlate well with the 78 wt%, 10 wt% and 11 wt% carbon, oxygen and
341 hydrogen reported for the HL biocrude in Table 4. Thus, while a wide range of compounds are
342 present in the HL biocrude, as a composite they are characteristic of a slightly unsaturated free
343 fatty acid, which in turn is due to the high concentration of lipid in the starting HL algae and
344 subsequent biocrude. Lipid extraction techniques would be expected to produce TAGs or FFAs
345 with similar concentrations of C, H and O. Hence, HTL serves as a biomass conversion and lipid
346 extraction process. As algal feedstocks increase substantially in lipid content, the biocrude from
347 HTL approaches the elemental makeup of a lipid extracted oil. Furthermore, the elevated TAN
348 number of the HL biocrude is consistent with a high concentration of FFAs produced from
349 hydrolysis of the initial lipids.

350 Selected compounds quantified via HPLC in the aqueous phase are also reported in Table
351 4. The selected compounds were the major aqueous phase organics observed and are consistent
352 with those observed using a saltwater *Nannochloropsis sp.* algal strain.[1] Notable increases in
353 the concentration of glycolic acid and glycerol were observed in the HL HTL aqueous phase.
354 The increased glycerol content in the HL HTL aqueous phase is consistent with TAGs
355 hydrolyzed during the HTL process that left the glycerol backbone intact. The low molecular

356 weight, highly oxygenated glycerol reported to the aqueous phase while the free fatty acid acyl
 357 compounds reported to the organic biocrude.

358 **Table 4.** Properties of the biocrude (organic), aqueous, gas and solid phases produced by HTL
 359 processing of the HL and SL algal feed slurries.

	SL	HL
<i>HTL Biocrude</i>		
Carbon, wt%	81	78
Hydrogen, wt%	9.9	11.2
Oxygen, wt%	2.6*	10.1
Nitrogen, wt%	5.7	0.5
Sulfur, wt%	0.4	0.1
Phosphorus, wt%	0.26	0.19
H:C molar ratio	1.46	1.71
TAN, mg _{KOH} /g _{oil}	60	180
Density, g/ml	0.97	0.91
Viscosity, cSt@40°C	178	40
Moisture, wt%	4.8	1.4
Ash, wt%	0.61	0.01
Filterable Solids, wt%	0.06	0.05
<i>HTL Aqueous</i>		
Nitrogen, wt. %	0.98	<0.05
Ammonia, wt. %	0.67	0.01
Carbon, wt. %	2.3	1.4
Total organic carbon, wt. %	2.2	1.5
Chemical oxygen demand, mg O/L	65,800	47,900
Chloride, ppm	19	21
Sulfur, ppm	190	200
pH	6.6	3.90
Selected compounds quantified by HPLC		
Glycolic acid, wt. %	0.03	0.39
Glycerol, wt. %	0.17	1.33
Acetic acid, wt. %	0.27	0.36
Propanoic acid, wt. %	0.17	0.07

Methanol, wt. %	0.11	0.2
Ethanol, wt. %	0.06	0.16

HTL Solids

Carbon, wt%	35	58
Hydrogen, wt%	5.3	8.0
Oxygen, wt%	21	15
Nitrogen, wt%	3.1	1.7
Sulfur, wt%	0.4	0.3
Phosphorus, wt%	34	17

HTL Gas Phase

Carbon dioxide, vol. %	100%	100%
------------------------	------	------

360 * Oxygen by difference was calculated at 4.5 wt% - a value consistent with similar biocrudes.

361 3.2 *HTL Biocrude Hydrotreating*

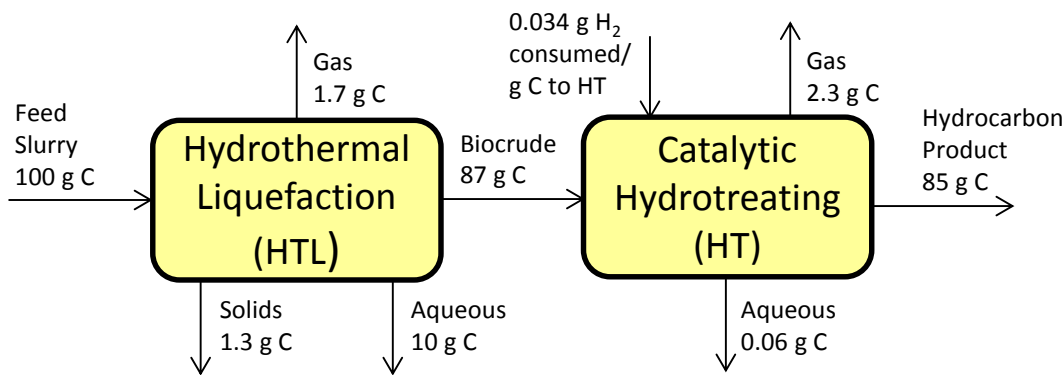
362 Both the HL and SL HTL biocrude phases were separated from their respective HTL
363 aqueous phases and upgraded via hydrotreating. The goal of hydrotreating was to remove
364 heteroatoms such as oxygen, nitrogen and sulfur in addition to increasing the H:C molar ratio of
365 the organic product. Both the HL and SL biocrudes were hydrotreated at 400°C and 1500 psig.
366 The liquid hourly space velocity (LHSV) of the feedrate to the reactor was held constant at 0.20
367 ml feed/ml-catalyst/h (or h⁻¹). However, as reported in Table 5, the dry-basis weight hourly
368 space velocity (WHSV) feed rate was slightly higher for the HL biocrude because it contained
369 less water.

370 The overall process block flow diagrams for each feedstock type on a normalized carbon
371 basis are illustrated in Figure 2. For the HL feedstock, 85% of the initial carbon was in the
372 hydrocarbon product. In contrast, the hydrocarbon product from the SL process contained only
373 54% of the carbon in the feed. Significantly more carbon was lost to the aqueous and gas phases

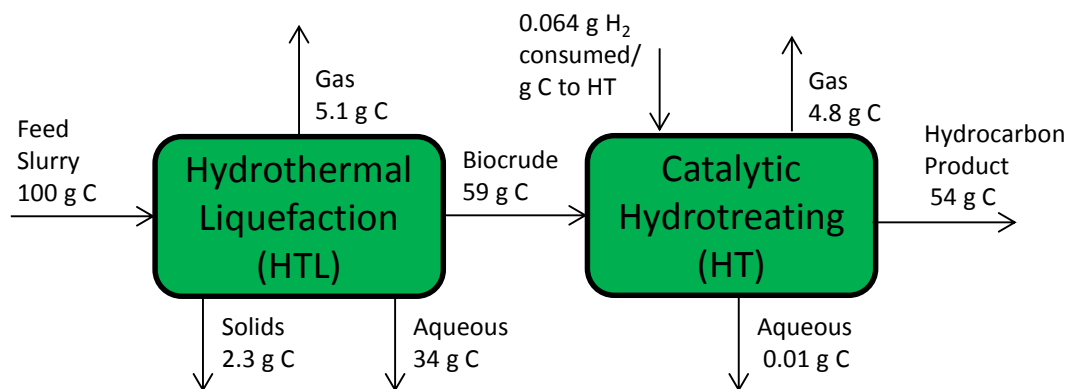
374 during HTL processing with the phototrophic feed. The significant (87%) carbon yield to
375 biocrude from the heterotrophic feedstock was due to the high concentration of lipids present.
376 The acyl lipid chains remain intact and preferentially report to the organic biocrude. In contrast,
377 the higher concentrations of carbohydrates and proteins in the SL algae produce more lower
378 molecular weight oxygenates, which contributes to greater loss of carbon to the aqueous and gas
379 phases

380 The hydrogen consumption observed when hydrotreating the HL biocrude was about a
381 factor of two lower than the SL biocrude on both a carbon (Figure 2) and (Table 5) mass basis.
382 The lower molecular weight oxygenates present in the SL biocrude may also contribute to higher
383 hydrogen consumption observed during HT. The HL biocrude begins with a higher H:C atomic
384 ratio than the SL biocrude. As shown in Table 6, the H:C ratio of the upgraded oils finishes at
385 the same value (2.0), suggesting less H₂ is required overall to increase the quality of the HL
386 organic phase.

387



Heterotrophic (HL) Feed



Phototrophic (SL) Feed

388

389 **Figure 2.** Block flow diagrams incorporating the HTL and HT process steps with inputs and
 390 outputs on a feed carbon basis for the heterotrophic and phototrophic feedstocks.

391

392 Another contributing factor to the lower hydrogen consumption of the HL biocrude is due
 393 in part to the diminished concentration of nitrogen present in the HL biocrude. The effect of
 394 nitrogen reduction during hydrotreating can be observed in the partitioning of the nitrogen to the
 395 HT products. The nitrogen in the SL biocrude was largely converted to ammonia, which
 396 accounted for a significant portion of the increased gas yield observed with the SL biocrude
 397 versus the HL biocrude. The ratio of methane:ethane in the HT gas product was greater for the

398 HL biocrude. This may suggest the catalyst was more active for cracking during the portion of
 399 the run when the HL biocrude was processed. The significant amount of ammonia produced
 400 during hydrotreating of the SL biocrude may have poisoned Brönsted acid sites on the
 401 hydrotreater catalyst, resulting in diminished hydrocarbon cracking activity when processing the
 402 SL biocrude. On the other hand, the SL biocrude produced significantly more hydrocarbon gas
 403 overall. Along with the greater biocrude nitrogen content, it is likely the increased concentration
 404 of low molecular weight compounds in the SL biocrude accounted for the overall increased gas
 405 yield versus the HL biocrude.

406

407 **Table 5.** Hydrogen consumption and phase yields observed during hydrotreating of the SL and
 408 HL HTL biocrudes at 400°C, 1500 psig and 0.20 mL wet biocrude/mL-catalyst/hr.

	SL	HL
WHSV, g biocrude (dry basis)/g catalyst/h	0.28	0.29
Hydrogen consumption, g H ₂ /g dry biocrude	0.051	0.026
Total Mass Balance, %	109	103
Hydrotreated organic mass yield, g/g dry biocrude	0.86	0.90
Hydrotreated organic volumetric yield, ml/ml wet biocrude	0.97	1.04
Aqueous yield, g/g dry biocrude	0.11	0.12
Gas yield, g/g dry biocrude	0.13	0.03

409

410

411

412

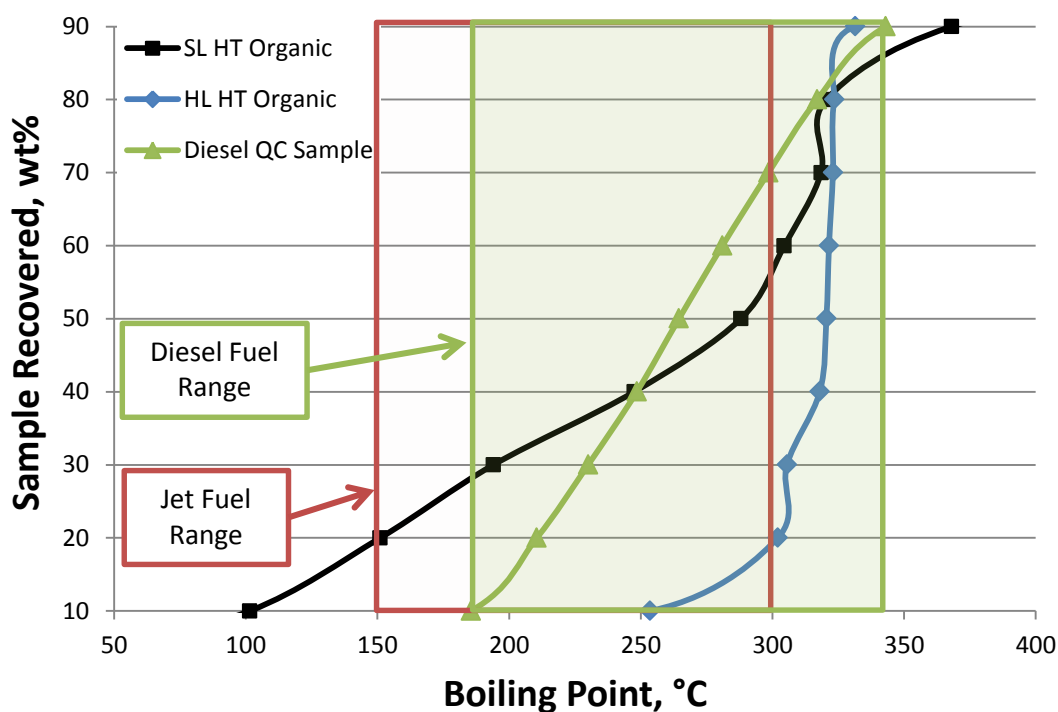
413 **Table 6.** Properties of the organic, aqueous and gas phases produced from hydrotreater
 414 processing of the HL and SL HTL biocrudes.

	SL	HL
<i>HT Organic Product</i>		
C, wt.% dry	84.2	86.0
H, wt.% dry	13.9	14.6
O, wt.% dry	1.7	1.7
N, wt.% dry	0.1	<0.05
S, ppm	63	18
Density, g/ml @ 40 °C	0.7747	0.7757
Viscosity, cSt	1.96	3.41
Water (KF), wt.%*	<0.2	<0.02
H/C molar ratio (dry basis)	2.0	2.0
<i>HT Aqueous Product</i>		
C, wt.%	0.11	0.44
N, wt.%	1.05	2.6
<i>HT Gas, (H₂ free basis)</i>		
CH ₄ , vol%	21.3	66.5
Ethane, vol%	16.0	16.4
Propane, vol%	6.6	11.8
Butanes, vol%	3.5	3.6
Pentanes, vol%	1.6	1.5
NH ₃ , vol%	50.6	0.0
H ₂ S, vol%	0.4	0.2

415 *The HT organic samples were washed with deionized water and dried with silica gel and
 416 sodium sulfate prior to KF analysis.

417
 418 The greater concentration of low molecular weight compounds in the SL biocrude can be
 419 observed in the composition analysis of the HT organic products. In Figure 3, the broad tail in
 420 the simulated distillation (simdis) results of SL HT organic product between 100-250°C is

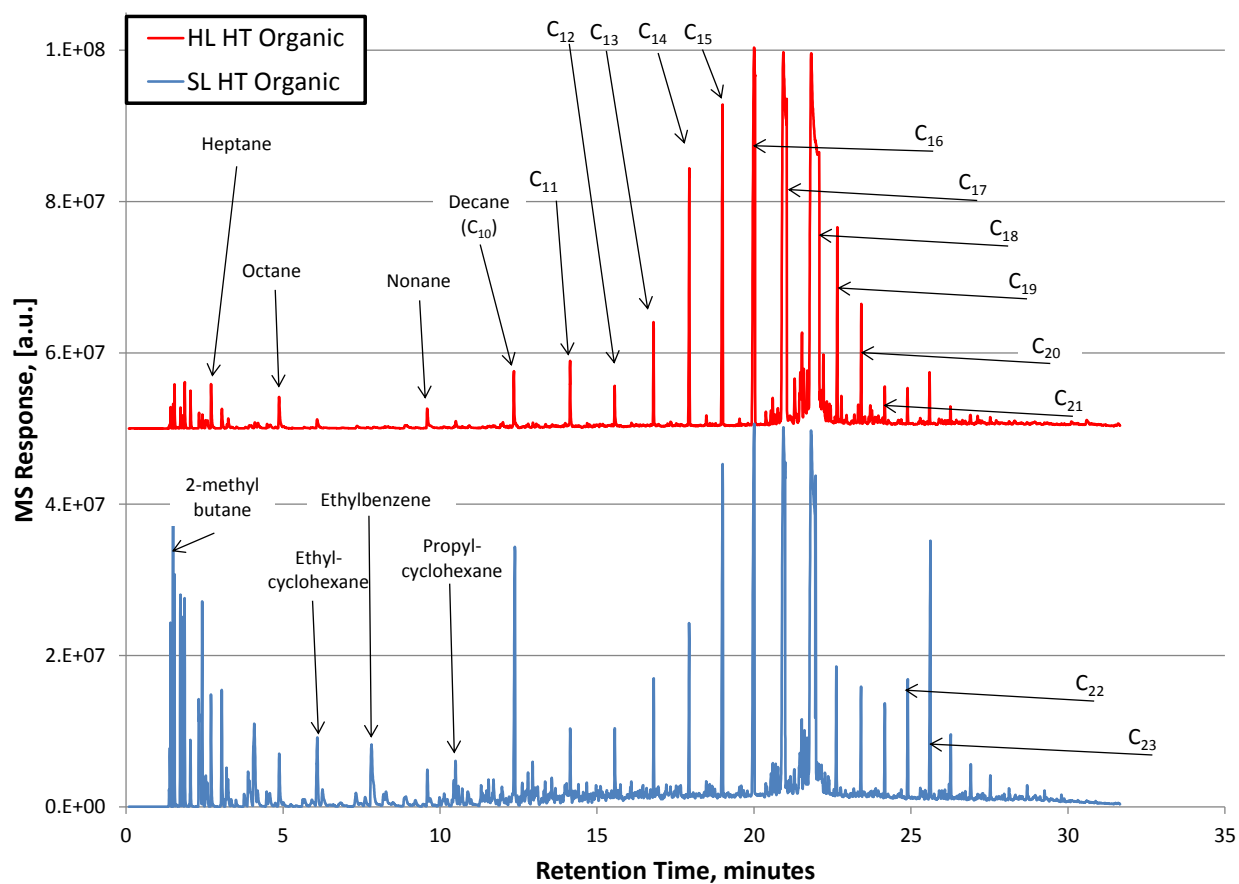
421 indicative of the lower boiling hydrocarbons. The HT hydrotreated oil contains a significantly
422 higher concentration of longer chain hydrocarbons derived from lipids. Thus, the HL HT
423 organic product boils within a fairly narrow range, all of which falls within the same range as the
424 diesel quality control (QC) sample. A portion of the SL sample has the potential to be a direct jet
425 fuel blendstock because a significant amount of its components boil in the jet range. Producing a
426 jet fuel fraction from the hydrotreated HL biocrude would require isomerization and possibly
427 hydrocracking in order to increase the concentration of compounds that boil in the jet fuel range.



428
429 **Figure 3.** Simulated distillation profiles of the HT organic fractions from the the HL and SL
430 feedstocks. The simulated distillation curve of a quality control (QC) diesel sample is also
431 included for comparison.
432

433 The greater concentrations of low molecular weight compounds in the SL sample can
434 also be observed the GC-MS trace of the HT organic samples (Figure 4). Significantly fewer
435 light compounds boiling below about 11 min retention time are observed in the HL product

436 versus the SL product. Odd numbered hydrocarbons are observed in the GC-MS traces in
437 addition to even numbered hydrocarbons. Only the even numbered hydrocarbons would be
438 expected biologically. The presence of odd numbered hydrocarbons demonstrates that
439 decarboxylation is an active pathway in the HT process. Decarboxylation is a mechanism
440 whereby a CO₂ is cleaved to produce a hydrocarbon with one carbon atom less than the original
441 acyl chain. Hydrodeoxygenation, whereby an even numbered hydrocarbon is produced with the
442 same number of carbons as the parent lipid acyl chain, is the other deoxygenation mechanism.



443

444 **Figure 4.** GC-MS profiles of the primarily hydrocarbon HT organic fractions from the the HL
445 and SL feedstocks.

446

447

448 The HT process was not optimized for this study. The LHSV of 0.20 hr^{-1} represents a
449 conservative feed rate. The goal of the hydrotreater test was to process the biocrude to a great
450 extent in order to compare the upgraded products. Higher LHSV's of 0.25 and 0.30 hr^{-1} were
451 tested with the HL biocrude and gave qualitatively similar results as the test conducted at 0.20 hr^{-1} .
452 The oxygen content of the HT organic phases processed at higher LHSV were within 12%
453 above and below those observed from processing at 0.20 hr^{-1} , suggesting the results were similar
454 within experimental error. Furthermore, the N content was reduced below the detection limit
455 ($<0.05 \text{ wt}\%$) at the higher space velocities and each sample had a similar H:C atomic ratio (2.0)
456 regardless of feed rate. Thus, the full activity of the catalyst was not challenged and a feed rate
457 at 0.30 hr^{-1} or greater could have been utilized before increases in oxygen and nitrogen
458 concentration were observed.

459 3.3 Whole Algae Conversion Techno-Economic Analysis

460 To explore the conversion cost implications of one strain cultivated in two different ways,
461 the experimental data from Sections 3.1 and 3.2 were used to develop commercial scale process
462 models and costs for the conversion steps. Figure 5 shows the block flow diagrams and
463 associated mass flow rates for the phototrophic SL feedstock and heterotrophic HL feedstock
464 conversion processes as modeled in AspenPlus. Note that only the conversion processes were
465 modeled; algae growth, harvesting and dewatering costs were not incorporated in the economic
466 analysis. The process steps inside the dashed line constitute the “conversion” process and were
467 the steps modeled and included in the TEA. The resulting HTL products (oil, solid, aqueous,
468 gas) are separated and the HTL oil is hydrotreated to form diesel and naphtha range fuels. The
469 HTL aqueous phase is catalytically treated via the Catalytic Hydrothermal Gasification process

470 to recover the carbon content and allow water recycle back to the growth ponds.[1] Process off
471 gas is supplemented with natural gas to generate hydrogen, heat and power. Nutrient recovery is
472 accomplished by recycling treated water, carbon dioxide containing flue gas, and treated solids
473 back to the algae ponds. Greater information on the methodology employed for the TEA as well
474 as model assumptions for the catalytic hydrothermal gasification and the hydrogen plant can be
475 found in Jones *et al.*[12]

476

477

478

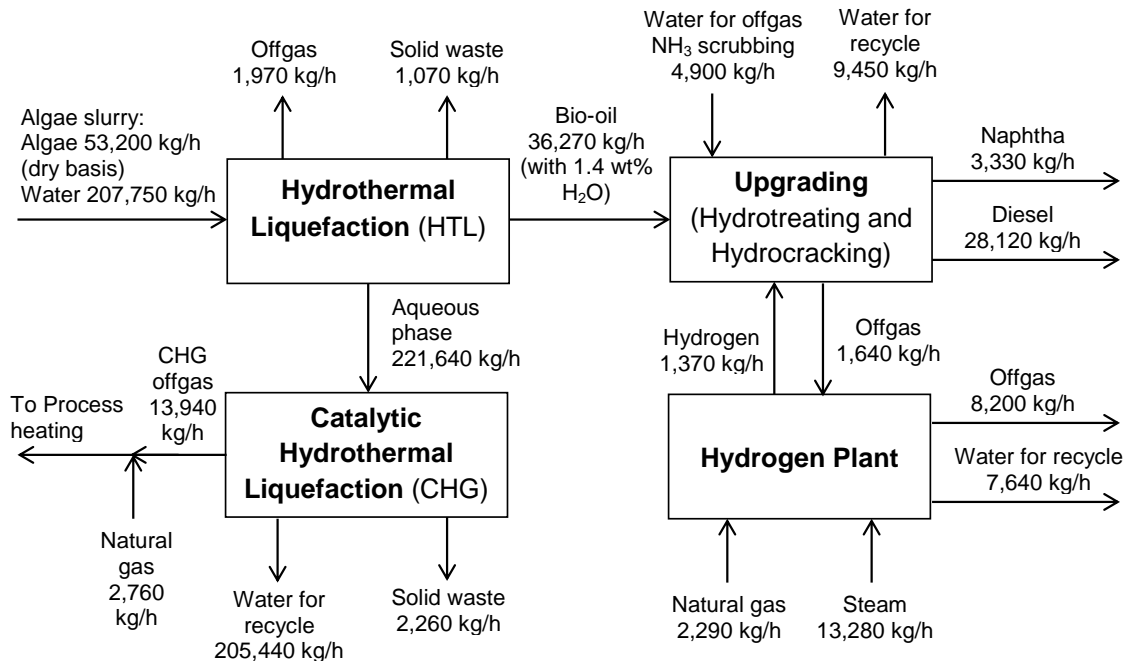
479

480

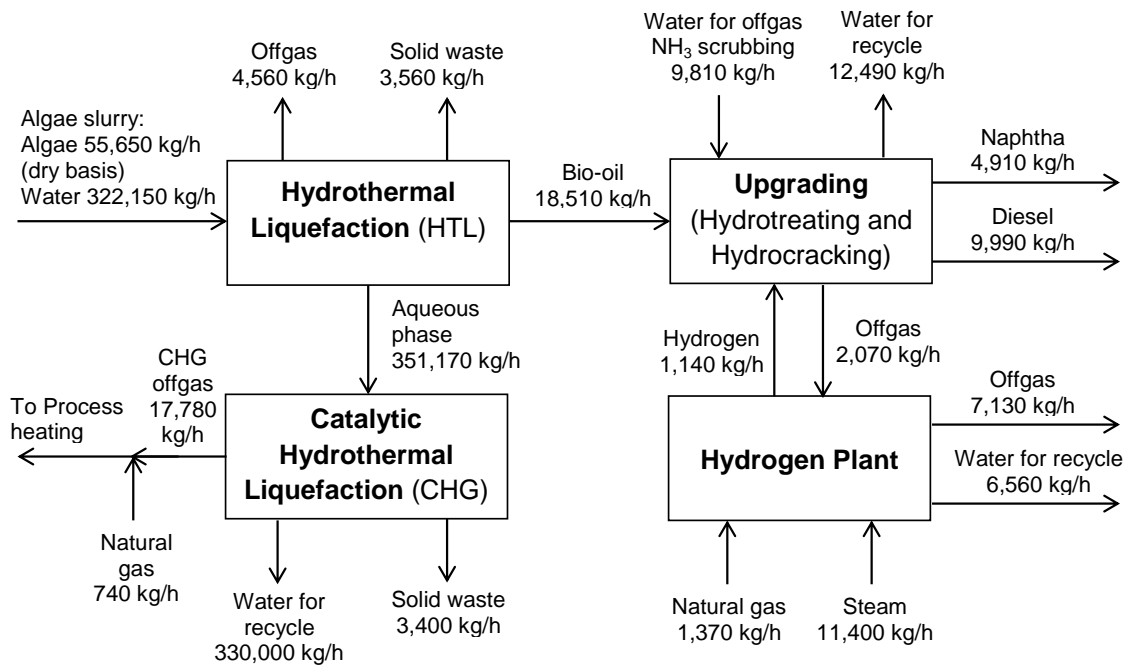
481

482

483



Heterotrophic (HL) Feed



Phototrophic (SL) Feed

Figure 5. Process block flow diagrams with mass flow rates for the conversion of whole algae to liquid transportation fuels as modeled in Aspen Plus.

492 Key modeled performance results are shown in Table 7. The assumed scale is 1215
493 metric tons per day of algae on a dry and ash free basis, consistent with Jones *et al.*[12] Thus the
494 algae slurry feed water flow rates differ for the two cases, reflecting the difference in the whole
495 algae ash content.
496

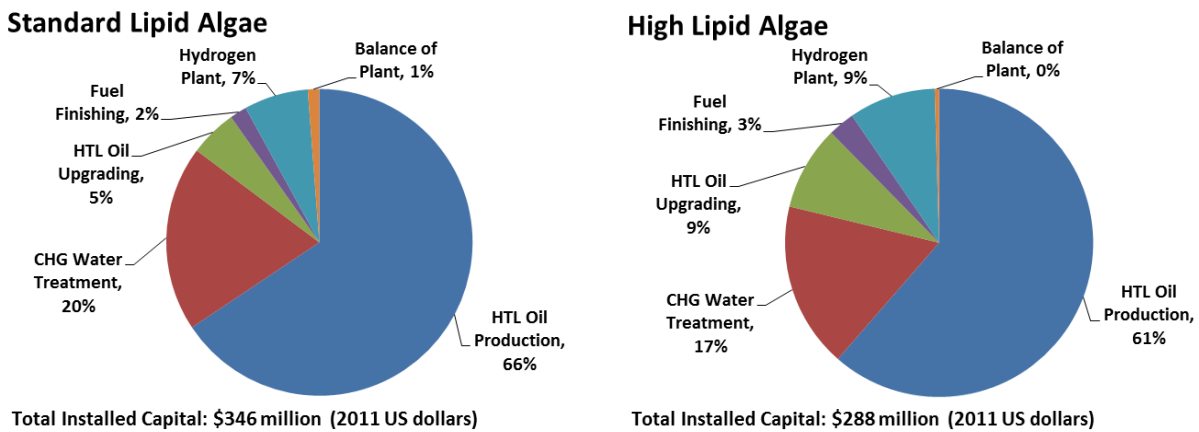
497 **Table 7.** Modeled performance results of the combined HTL and HT process to convert whole
 498 algae to liquid transportation fuels based on 1215 metric tons/day (dry and ash free basis).

	SL	HL	Ratio of HL/SL
<i>Raw Materials</i>			
Water in algae slurry feed, metric ton/h	322	207	0.6
Natural gas feed rate, metric ton/h	2.1	5.1	2.4
<i>Production Rates and Yields</i>			
Diesel production, million gallon gasoline equivalent (GGE)/yr	28	80	2.9
Byproduct (naphtha) production, million GGE/yr	14	9.4	0.7
Diesel yield, GGE/metric ton dry, ash-free algae	70	198	2.8
Byproduct (naphtha) yield, GGE/metric ton dry, ash-free algae	34	24	0.7
<i>Carbon Efficiency</i>			
Diesel and naphtha, % C from algae	49 %	79%	1.6
Diesel and naphtha, % C from algae and natural gas	44 %	69%	1.6
<i>Water Recycle</i>			
Water available for recycle to ponds, gal/gal total fuel	27	6.0	0.2
<i>Utilities</i>			
Electricity consumption, MW _e	10.9	9.0	1.2
Electricity generation, MW _e	2.6	2.0	0.8
Net electricity requirement, MW _e	8.3	7.1	0.9
Net electricity requirement, kW _e /gal total fuel	2.5	0.75	0.3
<i>Energy input</i>			
Algal feedstock, GJ/h HHV basis	1,210	1,620	0.7
Natural gas, GJ/h HHV basis	115	280	0.3
Electricity, GJ/h	30	26	0.4
<i>Energy output</i>			
Diesel, GJ/h HHV basis	470	1330	0.4
Byproduct (naphtha) GJ/h HHV basis	230	160	1.5
Overall energy efficiency, % HHV basis	52	77	0.7
Efficiency, based on algal feedstock only, % HHV basis	58	92	0.6
Natural Gas, GJ/h HHV basis	115	280	0.3

500 The SL feedstock has higher ash content than the HL feed. Thus, for the same ash-free
501 algae flowrate, the water content of the SL algae slurry feed is higher in order to be consistent
502 with the experimental data shown in Table 2. This is an artifact of holding the ash free algae
503 throughput constant, which is necessary to put the resulting yields on a comparable basis. The
504 total fuel yield (diesel plus naphtha) for the HL case is almost twice that for the SL algae case,
505 which is consistent with the biocrude yield experimental results shown in Table 3. This is
506 expected, given that that total lipid content is a key driver for biocrude yield. Natural gas is used
507 as makeup feed to the hydrogen plant. The natural gas usage for the HL algae case is greater
508 than that for the SL algae case for several reasons. Much of the feed to the hydrogen plant comes
509 from catalytically converting the organic material in the HTL aqueous phase to a medium
510 heating value gas (mostly methane and carbon dioxide). The HTL aqueous phase for the SL case
511 contains significantly more organic carbon than the HL case (Table 4). Additionally,
512 hydrotreating SL biocrude produces more hydrocarbon rich off gas (Table 6) that can be sent to
513 the hydrogen plant. Lastly, the yield of SL biocrude is lower, resulting in a lower hydrogen
514 demand. The SL case potentially recycles more water to ponds because of the higher water
515 content in the algae slurry feed. The SL feedstock has higher electricity consumption for HTL
516 and CHG processes because of the lower dry mass content (see Table 2) or higher water content
517 in the feed slurry than that of the high lipid one. With the same algae feed flow rates at dry ash
518 free basis for the two cases, higher water content in the feed stream leads to higher flow rate for
519 the wet feed slurry, which requires more electricity for process stream pumping. Therefore, the
520 net power requirement of the SL case is higher than the HL case.

521 Conversion installed capital costs are shown in Figure 6. The basis for the capital costs
522 can be found in Jones *et al.*[12] The HTL system is the largest fraction of the installed costs for

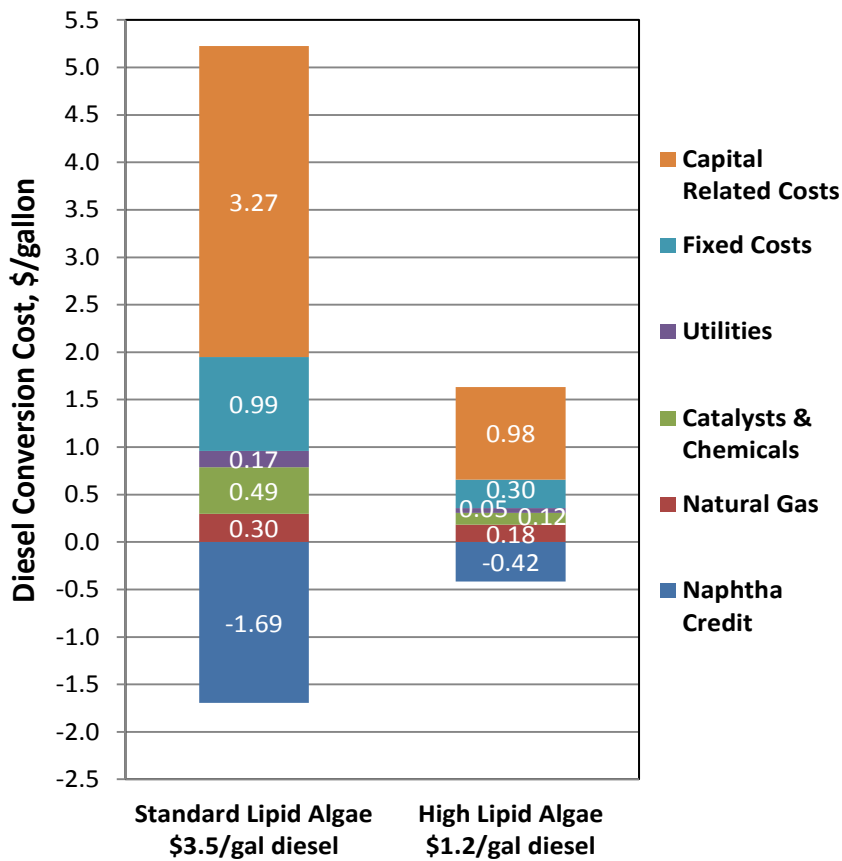
523 both cases: 66% for the SL and 61% for HL. The SL CHG costs are a greater proportion of the
 524 total versus the HL CHG. This is a function of both the HTL feed slurry dry mass concentration
 525 and biocrude yield. The SL algae case has a higher water concentration (i.e. a lower dry mass
 526 fraction) in the feed and a lower yield of biocrude. This in turn causes the SL case to have
 527 higher capital costs for the HTL and CHG processes and thus a higher total installed capital cost
 528 compared with the HL case. The higher water fraction of the feed also leads to a higher HTL
 529 aqueous phase product flow rate, which leads to larger CHG equipment and thus higher cost in
 530 the SL algae case. The HTL oil upgrading cost for the SL case is lower because of the lower bio-
 531 oil yield. This also results in a smaller, less expensive hydrogen plant. Overall, the HL case
 532 installed capital investment is about 17% lower than the SL case, primarily resulting from the
 533 higher dry mass fraction in the algae slurry feed.



535 **Figure 6.** Model results showing the relative contribution to the total installed capital costs for
 536 standard lipid (phototrophic) and high lipid (heterotrophic) algal feedstocks. The installed
 537 capital costs include the HTL, HT and CHG processes for each feedstock.

538
 539 The conversion cost breakouts on a dollar per gallon diesel equivalent basis by category
 540 are shown in Figure 7. The naphtha co-product is valued at \$3.25/gallon in both cases. Since the

541 naphtha boiling range material is a greater proportion of the overall product for the SL case, it
 542 has a greater cost impact relative to the HL case which produces greater diesel boiling range
 543 material. The total production cost for the high lipid case is 66% lower for the HL case than for
 544 the SL case primarily because of the higher final product yield for the HL feedstock as well as
 545 the lower HTL and CHG capital costs compared to the SL feedstock.



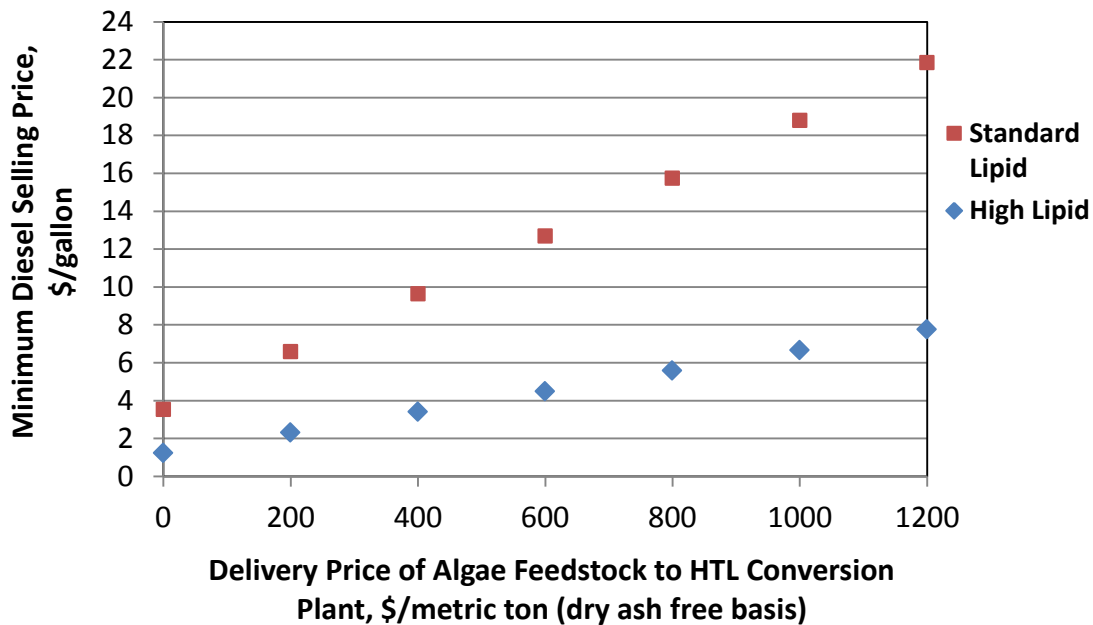
546
 547 **Figure 7.** Diesel conversion cost breakouts by contribution area on a per-gallon of diesel basis.
 548

549 Based on the above results and analysis, the dry mass content in the algae feed slurry is a
 550 major contributor to the overall conversion cost. To investigate the impacts of the dry mass
 551 content of the algae feed slurry on the cost, the dry mass fraction of the feed slurry for the SL

552 case was increased to the same as the HL case; other assumptions in the model remained
553 unchanged. Compared to the baseline SL case, the SL case with higher dry mass content resulted
554 in a 35% decrease in the natural gas requirement due to less process heating energy required in
555 the HTL and CHG processes. A 17% decrease in electricity consumption due to lower pumping
556 power was also observed in the model. Lower heat requirements also made more heat available
557 for steam generation and thus power generation. The net electricity generation of the new SL
558 case is about 130% of the baseline SL case. The overall energy efficiency increased about 2.5%
559 over the baseline case. Therefore, increasing the dry mass fraction in the feed slurry has a
560 moderate effect on the system performance.

561 The effect of the dry solids concentration in the SL feedstock on the processing cost was
562 also investigated. When the dry solids content of the SL feedstocks matched the HL feedstock,
563 the HTL process capital cost decreased from \$227 million to \$181 million, which is comparable
564 with the \$174 million of the HL case. The CHG process cost decreased about 21%, but was still
565 9% higher than the comparable HL case. The overall installed capital cost of the SL case with
566 greater solids concentration is about 17% lower than the baseline case and is comparable with
567 the HL case. Higher dry mass content in the feed slurry also leads to lower process heating
568 requirement for the HTL and CHG processes, which in turn leads to lower natural gas
569 requirement and thus lower variable operating cost. The diesel conversion cost of the SL case
570 with increased dry mass content in the feed is \$2.5/gallon, which is 28% lower than the baseline
571 SL case, though still much higher than the HL case at \$1.2/gallon. Therefore, the water
572 concentration in the feed has significant impact on the overall fuel production costs.

573 The opportunity cost related to the production of the algal feedstocks is assessed in
 574 Figure 8. The feedstock price for cultivating, harvesting and dewatering algae for delivery to the
 575 HTL plant was varied from \$0 to \$1200/dry ash-free metric ton. The minimum diesel selling
 576 price for the SL case ranges from nearly \$3.5 to \$22 per gallon of diesel. In contrast, for the same
 577 feedstock cost range, the HL related diesel selling price ranges from approximately \$1.2 to 7.7
 578 per gallon. The slope of the diesel conversion cost with the SL feedstock is about 36% greater
 579 than the HL. The primary driver for the difference in the slopes is the diesel fuel yield. The
 580 higher final product yield observed from the heterotrophically stressed HL feedstock
 581 significantly reduces the impact of feedstock price on the diesel fuel production cost.
 582



583
 584 **Figure 8.** Dewatered algae price delivered to the HTL plant and its impact on the minimum fuel
 585 conversion price.

586 Other reports of direct technical and economic comparisons of heterotrophically stressed
 587 versus phototrophically grown (but otherwise identical) unialgal feedstocks for the production of

588 fuels could not be located. Heterotrophic stressing could add significant cost to the production of
589 algal biomass due to the addition of a carbon substrate such as glucose. However, this study
590 presents new evidence that the improved hydrocarbon fuel yield obtained from heterotrophically
591 stressed algae cultures processed via HTL and HT can more than offset the associated costs
592 increase when compared directly to phototrophically grown algae. Hence, heterotrophic
593 stressing of algae cultures should be considered as a potentially more economically attractive
594 method to produce fuels in spite of increased production costs.

595 **4. Summary**

596 HTL processing produced twice the desired organic biocrude when the HL
597 heterotrophically grown algal material was converted versus the SL phototrophically grown
598 version of the same algal strain. The higher yield of biocrude was due to the substantially higher
599 lipid content of the HL feedstock. Upgrading of the biocrudes via HT produced similar yields on
600 an upgraded organic/g dry biocrude fed basis. Thus, the heterotrophically grown algal feedstock
601 produced about twice as much fuel blendstock. Furthermore, a greater portion of the upgraded
602 organic product from the HL feedstock contained hydrocarbons that boiled in the diesel fuel
603 range due to the high lipid content of the starting strain. The upgraded organic product from the
604 SL algae contained substantially more cyclic and lower-boiling compounds, likely due to the
605 higher concentration of proteins in the starting material. TEA modeling of a combined HTL and
606 subsequent HT process to convert the feedstocks to fuels revealed conversion costs associated
607 with the heterotrophic algae are lower compared to the phototrophic culture due to the greater
608 biocrude yields and hence greater yield to higher-value diesel. . The HTL capital costs

609 contributed the largest fraction in the total production cost for both feedstocks. . . For the first
610 time, we have demonstrated that although production costs will be higher when producing
611 heterotrophically stressed algae due to the addition of a carbon substrate, the significant
612 improvement of the fuel yield could more than offset the additional cost, resulting in a lower
613 minimum selling price for the fuel product.

614 **5. Acknowledgments**

615 The authors gratefully acknowledge the support for this research provided by the U.S.
616 Department of Energy through the Bioenergy Technologies Office (BETO). Pacific Northwest
617 National Laboratory is operated for the U.S. Department of Energy by Battelle under Contract
618 DE-AC06-76RL01830. We gratefully acknowledge Richard Lucke for running simulated
619 distillation and ppm S measurements on the hydrotreated organic products and Douglas Elliott
620 for helpful discussions and guidance provided during the preparation of this manuscript.

621 **6. References**

- 622 [1] D.C. Elliott, T.R. Hart, A.J. Schmidt, G.G. Neuenschwander, L.J. Rotness, M.V. Olarte, et al.
623 Process development for hydrothermal liquefaction of algae feedstocks in a continuous-flow
624 reactor. *Algal Research* 2 (2013) 445-54.
625
- 626 [2] P. Biller, A.B. Ross. Hydrothermal processing of algal biomass for the production of biofuels
627 and chemicals. *Biofuels* 3 (2012) 603-23.
628
- 629 [3] D. López Barreiro, W. Prins, F. Ronsse, W. Brilman. Hydrothermal liquefaction (HTL) of
630 microalgae for biofuel production: State of the art review and future prospects. *Biomass and*
631 *Bioenergy* 53 (2013) 113-27.
632
- 633 [4] C. Tian, B. Li, Z. Liu, Y. Zhang, H. Lu. Hydrothermal liquefaction for algal biorefinery: A
634 critical review. *Renewable and Sustainable Energy Reviews* 38 (2014) 933-50.

635
636 [5] D.C. Elliott, P. Biller, A.B. Ross, A.J. Schmidt, S.B. Jones. Hydrothermal liquefaction of
637 biomass: Developments from batch to continuous process. *Bioresource Technology* 178 (2015)
638 147-56.
639
640 [6] C. Jazrawi, P. Biller, A.B. Ross, A. Montoya, T. Maschmeyer, B.S. Haynes. Pilot plant
641 testing of continuous hydrothermal liquefaction of microalgae. *Algal Research* 2 (2013) 268-77.
642
643 [7] National Alliance for Advanced Biofuels and Bio-products (NAABB). Full Final Report,
644 Section II. 2013. p. 116.
645
646 [8] National Alliance for Advanced Biofuels and Bio-products (NAABB). Full Final Report,
647 Section II. 2013. p. 126-9.
648
649 [9] F. Delrue, Y. Li-Beisson, P.A. Setier, C. Sahut, A. Roubaud, A.K. Froment, et al.
650 Comparison of various microalgae liquid biofuel production pathways based on energetic,
651 economic and environmental criteria. *Bioresource Technology* 136 (2013) 205-12.
652
653 [10] L. Ou, R. Thilakaratne, R.C. Brown, M.M. Wright. Techno-economic analysis of
654 transportation fuels from defatted microalgae via hydrothermal liquefaction and
655 hydroprocessing. *Biomass and Bioenergy* 72 (2015) 45-54.
656
657 [11] Y. Zhu, K.O. Albrecht, D.C. Elliott, R.T. Hallen, S.B. Jones. Development of hydrothermal
658 liquefaction and upgrading technologies for lipid-extracted algae conversion to liquid fuels.
659 *Algal Research* 2 (2013) 455-64.
660
661 [12] S.B. Jones, Y. Zhu, D.B. Anderson, R.T. Hallen, D.C. Elliott, A.J. Schmidt, et al. Process
662 Design and Economics for the Conversion of Algal Biomass to Hydrocarbons: Whole Algae
663 Hydrothermal Liquefaction and Upgrading. Pacific Northwest National Laboratory, Richland,
664 WA., 2014. PNNL-23227. <http://www.osti.gov/scitech//servlets/purl/1126336/>
665
666 [13] R.E. Davis, D.B. Fishman, E.D. Frank, M.C. Johnson, S.B. Jones, C.M. Kinchin, et al.
667 Integrated Evaluation of Cost, Emissions, and Resource Potential for Algal Biofuels at the
668 National Scale. *Environmental Science & Technology* 48 (2014) 6035-42.
669
670 [14] Y. Zhu, M.J. Bidy, S.B. Jones, D.C. Elliott, A.J. Schmidt. Techno-economic analysis of
671 liquid fuel production from woody biomass via hydrothermal liquefaction (HTL) and upgrading.
672 *Applied Energy* 129 (2014) 384-94.
673
674 [15] D.C. Elliott, H. Wang, R. French, S. Deutch, K. Iisa. Hydrocarbon Liquid Production from
675 Biomass via Hot-Vapor-Filtered Fast Pyrolysis and Catalytic Hydroprocessing of the Bio-oil.
676 *Energy & Fuels* 28 (2014) 5909-17.
677
678 [16] N. Sudasinghe, B. Dungan, P. Lammers, K. Albrecht, D. Elliott, R. Hallen, et al. High
679 resolution FT-ICR mass spectral analysis of bio-oil and residual water soluble organics produced

680 by hydrothermal liquefaction of the marine microalga *Nannochloropsis salina*. *Fuel* 119 (2014)
681 47-56.
682
683